# UV Advanced Oxidation

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#### a place of mind

#### THE UNIVERSITY OF BRITISH COLUMBIA

### **Drinking Water**











#### Micropollutants in Water (Global Dimension)

#### Anthropogenic fluxes affecting water quality

	10 <sup>6</sup> tons /year
Global fertilizer production (2000)	140
Global pesticide production	5
Synthetic organic production	300
Oil spills (average 1980-2000)	0.4

source: Science Magazine vol. 313 (2006)

## **Drinking Water**

# Treatment necessary before consumption for:

- Disinfection (removal of pathogens)
- Reduction of disinfection by-product (DBP) formation
- Removal of emerging pollutants
- Improving taste and odour



### **UV** Disinfection

Presently

# Several thousands drinking water facilities use UV based disinfection



## Is the application of UV limited to water disinfection?

## Can UV play a role in addressing other water quality challenges?



#### **UV-Based Oxidation Processes**

UV photochemical oxidation processes for water treatment involve:

- Direct photolytic action of UV on dissolved matter in water (e.g. micropollutants)
- Photochemically assisted production of oxidants for removing harmful organic matter
- Photochemically assisted catalytic processes



#### **UV-based Oxidation**

- 1879 H<sub>2</sub>O<sub>2</sub> decomposition by sunlight (Downes & Blunt)
- 1899 Photolysis of carbonic acid (Bach)
- 1907 H<sub>2</sub>O<sub>2</sub> decomposition by UV (Thiele)
- 1922 H<sub>2</sub>O<sub>2</sub> photolysis and reaction products (Kornfeld)
- 1936 Photomineralization by VUV (Fricke & Hart)
- 1971 Photocatalytic oxidation of single compounds in the gas phase
- 2004 First full scale UV-H2O2 plant for water treatment

#### UV Photooxidation Advanced Oxidation Process (AOP)





#### OH Radical Oxidation Potential

Oxidizing species	Oxidation	
	potential (eV)	
Fluorine	3.06	
Hydroxyl radical	2.80	
Atomic oxygen	2.42	
Ozone	2.07	
Hydrogen peroxide	1.78	
Chlorine dioxide	1.57	
Chlorine	1.36	
Oxygen	1.23	



#### **UV Oxidation Research in Water**

#### <u>To Date:</u> Worldwide: ~5500



#### UV Oxidation for Water Purification Patents

To Date:

Number of Patents Worldwide: ~1,100



## **UV AOP in Drinking Water**

- Ultraviolet + Hydrogen Peroxide: UV/H<sub>2</sub>O<sub>2</sub>
- Photocatalysis: UV/TiO<sub>2</sub>
- Vacuum Ultraviolet: VUV



#### **Target Contaminants**

- Micro-pollutants
  - Pesticides
  - Pharmaceuticals
  - Personal care products
- Taste & odor (T&O)
- Gasoline additives
- Surfactants
- Flame retardants





UV-H<sub>2</sub>O<sub>2</sub> Advanced Oxidation

![](_page_14_Picture_1.jpeg)

# UV-H<sub>2</sub>O<sub>2</sub> AOP

 $\rm H_2O_2 + hv_{<270~nm} \rightarrow 2HO \bullet$ 

![](_page_15_Figure_2.jpeg)

![](_page_15_Picture_3.jpeg)

#### Impact of UV/H<sub>2</sub>O<sub>2</sub> on Pesticides

![](_page_16_Figure_1.jpeg)

#### Impact of UV/H<sub>2</sub>O<sub>2</sub> on EDCs

![](_page_17_Figure_1.jpeg)

![](_page_17_Picture_2.jpeg)

#### **Drinking Water UV-H<sub>2</sub>O<sub>2</sub> AOP**

#### **Treatment conditions (commercial applications)**

- Fluence:  $\geq 600 \text{ mJ/cm}^2$
- H<sub>2</sub>O<sub>2</sub> dose: 5 to 15 mg/L
- LP and MP Hg lamp technology

![](_page_18_Picture_5.jpeg)

#### **Drinking Water UV-H<sub>2</sub>O<sub>2</sub> AOP**

#### **Commercial drinking water applications**

Installation	UV System	<b>Target Pollutant</b>
Orange County Water District, CA, USA	TrojanUVPhox™	NDMA; 1,4-dioxane
West Basin Municipal Water District, CA, USA	TrojanUVPhox™	NDMA
Stockton, CA, USA	TrojanUVPhox™	1,4-dioxane
PWN Treatment Plant Andijk, Netherlands	TrojanUVSwift™ ECT	Pesticides
City of Cornwall, ON, Canada	TrojanUVSwift™ ECT	MIB & geosmin
Salt Lake City Department of Public Utilities, UT, USA	Rayox™	PCE

Sarathy, S.R. and Mohseni, M. 2006. An overview of UV-based advanced oxidation processes for drinking water treatment. IUVA News. 8(2): 16-27.

![](_page_19_Picture_4.jpeg)

![](_page_20_Picture_0.jpeg)

![](_page_21_Picture_0.jpeg)

# UV Photocatalysis

![](_page_22_Picture_1.jpeg)

![](_page_23_Figure_0.jpeg)

#### **UV Photocatalysis**

![](_page_24_Figure_1.jpeg)

![](_page_24_Picture_2.jpeg)

### **TiO**<sub>2</sub> **Photocatalyst**

# Well known photocatalyst V Chemically stable

Strong oxidation-reduction reactions

![](_page_25_Picture_5.jpeg)

**Commercial TiO**<sub>2</sub> (Degussa P-25) 80% Anatase – 20% Rutile

#### PCO of Pharmaceuticals with suspended TiO<sub>2</sub>

![](_page_26_Figure_1.jpeg)

#### **Photocatalytic Reactors**

# State of the photocatalyst

1. Slurry

2. Immobilized

3. Fluidized bed

![](_page_27_Figure_5.jpeg)

#### **Slurry vs. Fluidized Bed PCO**

![](_page_28_Figure_1.jpeg)

## **Slurry vs. Fluidized Bed PCO**

Comparable degradation of contaminant in slurry and fluidized bed photocatalytic reactor

No need for downstream filtration • More feasible with fluidized bed • • process process

![](_page_29_Picture_3.jpeg)

#### PCO of 2,4-D (fluidized bed photoreactor)

![](_page_30_Figure_1.jpeg)

# Vacuum UV (VUV) Photooxidation

![](_page_31_Picture_1.jpeg)

## VUV

- Radiation with wavelength between 10 200 nm
- VUV photolyzes water molecules, which produce HO• radicals

High absorptivity of water at low wavelengths

 $H_2O + hv_{vuv} \rightarrow HO^{\bullet} + H^{\bullet}$  $H_2O + hv_{vuv} \rightarrow HO^{\bullet} + e^- + H^+$ 

- Two common sources of VUV
  - ✓ Excimer lamp: 173 nm
  - ✓ Hg vapor lamp: 185 nm (+ 254 nm)

![](_page_32_Picture_8.jpeg)

![](_page_33_Figure_0.jpeg)

#### **VUV Photooxidation of T&O**

![](_page_34_Figure_1.jpeg)

#### **Comparison of Different AOPs**

![](_page_35_Figure_1.jpeg)

Imoberdorf & Mohseni, 2011

![](_page_35_Picture_3.jpeg)

Large Scale Applications

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- There have been more than 100 full scale UV-based AOP installations around the world
- UV-H<sub>2</sub>O<sub>2</sub> is the most applied technology in large scale
- Often applied to cases (i.e. contaminants) which show persistence towards ozone and/or cases where formation of bromate in bromide containing source waters is a concern

![](_page_37_Picture_4.jpeg)

![](_page_38_Picture_1.jpeg)

Surface water

 (IJssel Lake) – as the
 source of drinking
 water to North Holland area

![](_page_38_Figure_3.jpeg)

Conventional water treatment system commissioned nearly 50 years ago

![](_page_38_Picture_5.jpeg)

- Taste & Odour problem has been a concern
- GAC was installed about 30 years to address the problem

![](_page_39_Figure_3.jpeg)

**Issues and Challenges with Conventional Treatment** 

- Chlorine, as primary disinfectant, was not effective against parasites (Giardia and Cryptosporidium)
- Chlorine also led to DBP formation in the distributions system
- Very small levels of nitrate, pesticides, pharmaceuticals were being detected due to increased industrials activities
- Bromate formation made ozone unfeasible

![](_page_40_Picture_6.jpeg)

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![](_page_41_Picture_6.jpeg)

UV-H2O2 AOP was considered as a viable alternative

- It was investigated against a range of contaminants (e.g., atrazine, NDMA, MTBE, 1,4dioxane, bisphenol A, microcystine and pharmaceuticals e.g., diclofenac, ibuprofen)
- Studies focused primarily on medium pressure lamp UV system
- The use of UV also provided primary disinfection, helping to eliminate the use of Cl<sub>2</sub>

![](_page_42_Picture_5.jpeg)

#### **Full Scale UV-H**<sub>2</sub>**O**<sub>2</sub> *Andijk – the Netherlands* Inactivation of 3 different pathogens

![](_page_43_Figure_1.jpeg)

#### **Full Scale UV-H<sub>2</sub>O<sub>2</sub>** *Andijk – the Netherlands* Removal of selected micropollutants

![](_page_44_Figure_1.jpeg)

The technology was adopted and a full scale system was installed in 2004, treating a design flow rate of about 4000 m<sup>3</sup>/h and 80% destruction of atrazine

 All contaminants were removed at UV doses up to about 540 mJ/cm<sup>2</sup> (0.56 kWh/m<sup>3</sup>) and hydrogen peroxide concentrations of about 6 mg/L

![](_page_46_Figure_1.jpeg)

#### **Selected Full Scale UV-H<sub>2</sub>O<sub>2</sub>**

Location	Contaminants	Flow rate (MLD)	Technology used
Region of Peel, ON, Canada	MIB, Geosmin	390	UV-H2O2 (MP UV)
Orange County,	NDMA,	379	UV-H2O2 (LP
CA, USA	1,4-Dioxane		UV)
PWN Heemskerk, Netherlands	Pesticides	144	UV-H2O2 (MP UV)
Aurora, Colorado,	NDMA,	190	UV-H2O2 (LP
USA	1,4-Dioxane		UV)
Luggage Point,	NDMA,	70	UV-H2O2 (LP
Brisbane	1,4-Dioxane		UV)

![](_page_47_Picture_2.jpeg)

#### Summary

UV based AOPs are rapidly growing solution alternatives for:

- Degradation of micro-pollutants in drinking water sources
  - In addition to **DISINFECTION**
- Removal of taste and odor (T&O) from drinking water
- Water re-use and reclamation

![](_page_48_Picture_6.jpeg)

#### Thank You !

![](_page_49_Picture_1.jpeg)

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![](_page_50_Picture_0.jpeg)

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